

I. Schwefelsäure.

Tabelle A.

Die Fehler in δ :

Concentration	$\Delta \delta =$
0.9	$0.56 \Delta v_1 - 1.67 \Delta v_2 + 5.49 \Delta d_1 + 0.61 \Delta d_2 - 5.92 \Delta D$
0.8	$0.68 \Delta v_1 - 1.24 \Delta v_2 + 2.83 \Delta d_1 + 0.71 \Delta d_2 - 3.38 \Delta D$
0.7	$0.83 \Delta v_1 - 1.08 \Delta v_2 + 1.97 \Delta d_1 + 0.85 \Delta d_2 - 2.67 \Delta D$
0.6	$1.06 \Delta v_1 - 0.91 \Delta v_2 + 1.55 \Delta d_1 + 1.03 \Delta d_2 - 2.41 \Delta D$
0.5	$1.38 \Delta v_1 - 0.86 \Delta v_2 + 1.31 \Delta d_1 + 1.31 \Delta d_2 - 2.45 \Delta D$
0.4	$1.87 \Delta v_1 - 0.82 \Delta v_2 + 1.17 \Delta d_1 + 1.75 \Delta d_2 - 2.74 \Delta D$
0.3	$2.70 \Delta v_1 - 0.82 \Delta v_2 + 1.08 \Delta d_1 + 2.52 \Delta d_2 - 3.43 \Delta D$
0.2	$4.47 \Delta v_1 - 0.81 \Delta v_2 + 1.03 \Delta d_1 + 4.10 \Delta d_2 - 4.93 \Delta D$
0.1	$9.75 \Delta v_1 - 0.83 \Delta v_2 + 1.00 \Delta d_1 + 9.00 \Delta d_2 - 9.82 \Delta D$